

## Supporting Information © Wiley-VCH 2013

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## Electric Current Test Paper Based on Conjugated Polymers and Aligned Carbon Nanotubes\*\*

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## **Supporting Information**

## **Experimental Section**

Trimethyl[4-(phenylethynyl)phenyl]-silane (p-Me<sub>3</sub>SiDPA, 99%) was obtained from Alfa Aesar, and the other chemicals were ordered from Sinopharm Chemical Reagent Co., Ltd. Aligned multi-walled carbon nanotube (MWCNT) sheets with lengths of up to meters were spun from a MWCNT array, which was synthesized by a chemical vapor deposition. The growth conditions of MWCNT arrays were summarized as below: thickness of 1.2 nm for Fe and thickness of 3 nm for Al<sub>2</sub>O<sub>3</sub> (the Fe and Al<sub>2</sub>O<sub>3</sub> catalyst system was coated on silicon substrate by electron beam evaporation with rates of 0.5 and 2 Å/s for Fe and Al<sub>2</sub>O<sub>3</sub>, respectively), flow rate of 400 standard cm<sup>3</sup>/min for argon, flow rate of 30 standard cm<sup>3</sup>/min for hydrogen, flow rate of 90 standard cm<sup>3</sup>/min for ethylene, growth temperature of 740 °C, and growth time of 10 min.

Mechanical tests were performed on a Hengyi Table-Top Universal Testing Instrument. The composite samples were mounted onto sample holders with a gauge length of 5 mm, and their cross-sectional sizes were determined by SEM. The measurement typically included three steps: (1) fixing two ends of a composite film or fiber onto the sample hold with silver paste; (2) cutting the two shoulders of the sample holder; (3) starting the testing procedure to obtain a stress-strain curve. Electrical conductivities were obtained by a two-probe method. The two ends of a composite sample were connected to two copper wires by conductive silver paste, and the copper wires were further connected to a digital Source-Meter for a measurement.

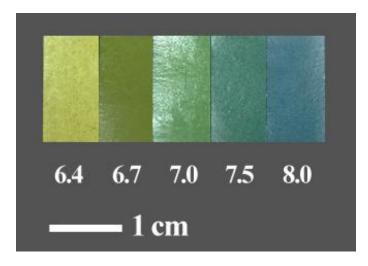
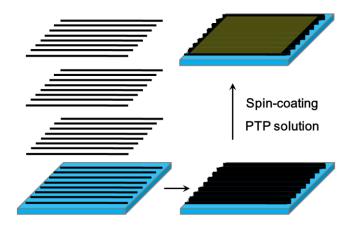
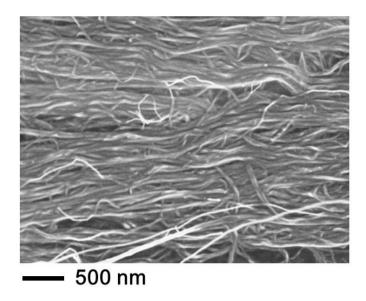


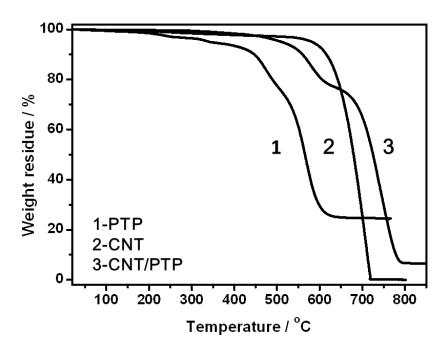
Figure S1. Photograph of colorimetric card of precision pH test paper.



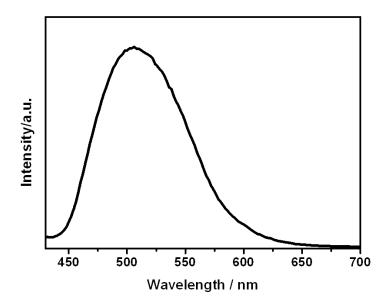
**Figure S2.** Schematic illustration to the preparation of an aligned MWCNT/PTP composite film by a spin-coating process.



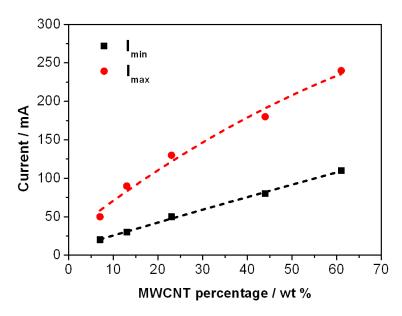
**Figure S3.** Scanning electron microscopy (SEM) image of an aligned MWCNT/PTP composite film.



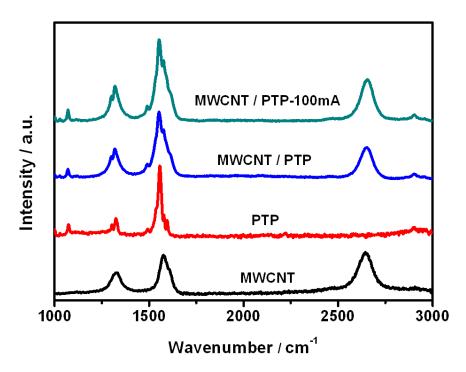
**Figure S4.** TGA curves of pristine MWCNTs, PTP and MWCNT/PTP composite film in air.



**Figure S5.** Typical fluorescent spectrum of a MWCNT/PTP composite film.



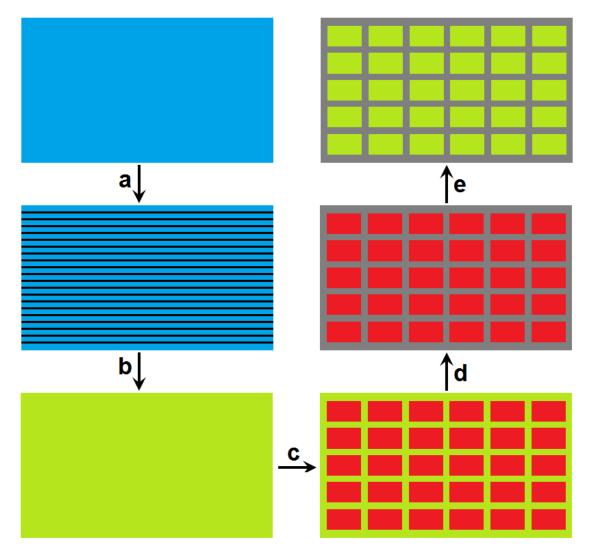
**Figure S6.** Dependence of the current range for the electrochromatic transition on the MWCNT weight percentage.



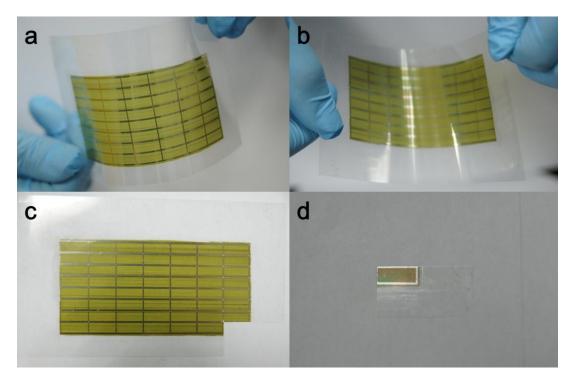
**Figure S7.** Raman spectra of bare MWCNT, bare PTP and MWCNT/PTP composite film before and after pass with a current of 100 mA.



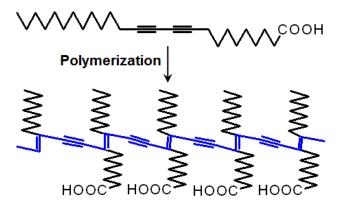
**Figure S8.** Fluorescent micrographs of a pure PTP film being heated to different temperatures.



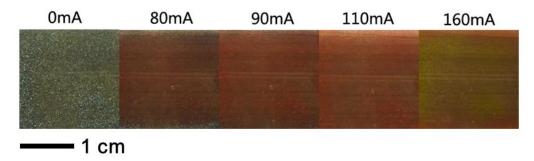
**Figure S9.** Schematic illustration to the fabrication of standard samples. **a.** Attaching aligned MWCNT sheet onto a substrate. **b.** Coating a PTP solution onto the MWCNT sheet. **c.** Covering a mask with rectangles that shared the same length and width and were separated with the same width on the composite film. **d.** Depositing a layer of metal that functions as electrodes for the standard samples. **e.** Removing the mask to produce standard samples.



**Figure S10. a** and **b.** Photographs of flexible MWCNT/PTP composite films (56 standard units) on poly (ethylene terephthalate) substrate. **c** and **d.** Photographs for one standard unit of a MWCNT/PTP composite film being cut from **a**.



**Figure S11.** Synthesis of the conjugated polymer by topochemical polymerization of  $CH_3(CH_2)_{11}CCCC(CH_2)_8COOH$  under UV light.



**Figure S12.** Photographs of an aligned MWCNT/polydiacetylene composite film when passed with increasing currents from 0 to 160 mA.